

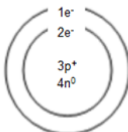
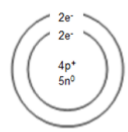
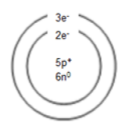
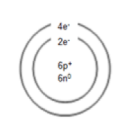
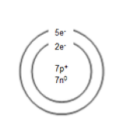



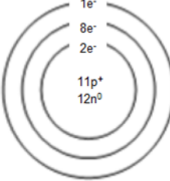
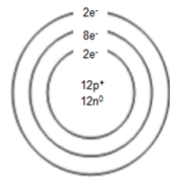
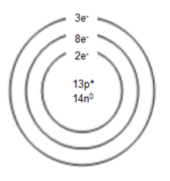
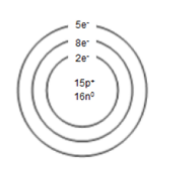
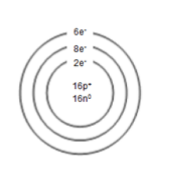
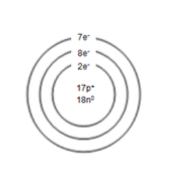
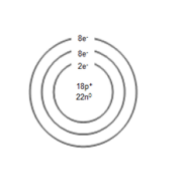
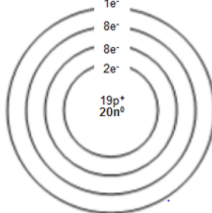
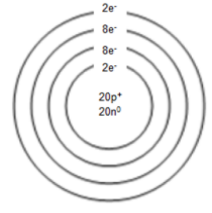
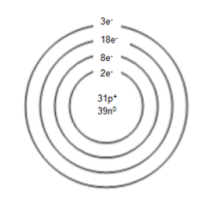
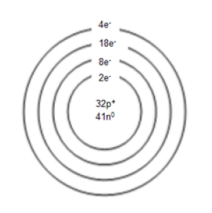
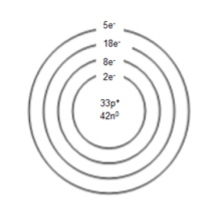
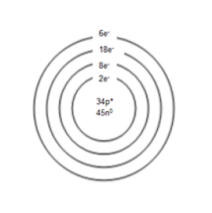



# Periodic Table Cards

<p><b>Hydrogen (H)</b></p>  <p>Ionization Energy: 1312 kJ/mol Electronegativity: 2.1</p>							<p><b>Helium (He)</b></p>  <p>Ionization Energy: 2372 kJ/mol Electronegativity: N/A</p>
<p><b>Lithium (Li)</b></p>  <p>Ionization Energy: 520 kJ/mol Electronegativity: 1.0</p>	<p><b>Beryllium (Be)</b></p>  <p>Ionization Energy: 900 kJ/mol Electronegativity: 1.5</p>	<p><b>Boron (B)</b></p>  <p>Ionization Energy: 801 kJ/mol Electronegativity: 2.0</p>	<p><b>Carbon (C)</b></p>  <p>Ionization Energy: 1086 kJ/mol Electronegativity: 2.5</p>	<p><b>Nitrogen (N)</b></p>  <p>Ionization Energy: 1402 kJ/mol Electronegativity: 3.0</p>	<p><b>Oxygen (O)</b></p>  <p>Ionization Energy: 1314 kJ/mol Electronegativity: 3.5</p>	<p><b>Fluorine (F)</b></p>  <p>Ionization Energy: 1681 kJ/mol Electronegativity: 4.0</p>	<p><b>Neon (Ne)</b></p>  <p>Ionization Energy: 2081 kJ/mol Electronegativity: N/A</p>
<p><b>Sodium (Na)</b></p>  <p>Ionization Energy: 496 kJ/mol Electronegativity: 0.9</p>	<p><b>Magnesium (Mg)</b></p>  <p>Ionization Energy: 738 kJ/mol Electronegativity: 1.2</p>	<p><b>Aluminum (Al)</b></p>  <p>Ionization Energy: 578 kJ/mol Electronegativity: 1.5</p>	<p><b>Missing Element #1</b></p>	<p><b>Phosphorus (P)</b></p>  <p>Ionization Energy: 1011 kJ/mol Electronegativity: 2.1</p>	<p><b>Sulfur (S)</b></p>  <p>Ionization Energy: 1000 kJ/mol Electronegativity: 2.5</p>	<p><b>Chlorine (Cl)</b></p>  <p>Ionization Energy: 1251 kJ/mol Electronegativity: 3.0</p>	<p><b>Argon (Ar)</b></p>  <p>Ionization Energy: 1521 kJ/mol Electronegativity: N/A</p>
<p><b>Potassium (K)</b></p>  <p>Ionization Energy: 419 kJ/mol Electronegativity: 0.8</p>	<p><b>Calcium (Ca)</b></p>  <p>Ionization Energy: 550 kJ/mol Electronegativity: 1.0</p>	<p><b>Gallium (Ga)</b></p>  <p>Ionization Energy: 558 kJ/mol Electronegativity: 1.7</p>	<p><b>Germanium (Ge)</b></p>  <p>Ionization Energy: 709 kJ/mol Electronegativity: 1.8</p>	<p><b>Arsenic (As)</b></p>  <p>Ionization Energy: 834 kJ/mol Electronegativity: 1.9</p>	<p><b>Selenium (Se)</b></p>  <p>Ionization Energy: 869 kJ/mol Electronegativity: 2.1</p>	<p><b>Missing Element #2</b></p>	<p><b>Krypton (Kr)</b></p>  <p>Ionization Energy: 1170 kJ/mol Electronegativity: N/A</p>